SHG PHASE MATCHING CONDITIONS FOR UNDOPED AND DOPED LITHIUM NIOBATE

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We measured the refractive indices of undoped and Zn- or Mg-doped lithium niobate in the wavelength range 400 to 1200 nm. The results are described by a generalized Sellmeier equation which takes into account the defect structure of the material. From the generalized Sellmeier equation a simple relation between the phase matching temperature for noncritical type I second harmonic generation (SHG) and the internal critical phase matching angle is derived. Good correspondence with experimental data over a wide composition and fundamental wavelength range is observed.

Key words: LiNbO3, refractive index, nonlinear effects.

1 INTRODUCTION

Lithium niobate can be fabricated with variable Li₂O contents (46–50 mol%) and additional dopants. For Mg and Zn doping an abrupt change of several physical properties at a certain doping level (threshold concentration) is observed. Since many optical applications depend on the refractive indices, a precise description of the refractive indices of LiNbO₃ as a function of the composition is of great importance. Here we use a Sellmeier equation with four independent variables—Li content, dopant concentration (Mg or Zn), wavelength and temperature —to derive a simple relation between the phase matching temperature for noncritical type I second harmonic generation and the phase matching angle.

2 EXPERIMENTAL

We measured the refractive indices of undoped and doped lithium niobate by an interferometric technique⁴ in a wavelength range from 400 to 1200 nm. Five samples were grown by the Czochralski technique from melts with varying Li content and were characterized by means of their Curie temperature.⁵ Stoichiometric material was prepared with K_2O in the melt⁶ or by the vapour transport equilibration (VTE) technique.⁷ The doped samples were grown from a congruent melt with up to 9 mol% XO (X = Mg, Zn). The Mg concentration in the crystals was determined applying literature data,⁸ the Zn content was measured using an electron microprobe.⁹

3 RESULTS

The refractive indices are excellently described by our generalized Sellmeier equation

$$T_{PM} = [(1067)^2 + 3*10^6 \cos^2\theta_{PM}]^{1/2} - 1042$$

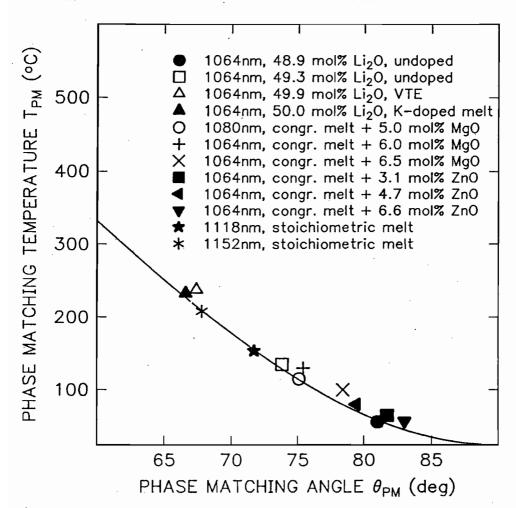


FIGURE 1 Phase matching temperature $T_{\rm PM}$ for noncritical type I second harmonic generation as a function of the internal phase matching angle $\theta_{\rm PM}$. The curve is derived from the generalized Sellmeier equation (see text). Data points represent our measurements and results reported by other authors. ^{13,14} Fundamental wavelength and crystal composition for each data point are listed in the inset.

which takes into account the defect structure of Li-deficient and Mg- or Zn-doped lithium niobate.

The equation consists of several oscillator terms representing the contribution from the NbO₆-octahedron $(A_{0,i}/(\lambda_{0,i}^{-2}-\lambda^{-2}))$ and approximated contributions from the IR and far UV region $(A_{\text{IR},i} \lambda^2 + A_{\text{UV}})$. The temperature dependence of the resonance wavelength is assumed to be proportional to the temperature dependence of the band gap $\lambda_{0,i}(T) = \lambda_{0,i} + \mu_{0,i}(f(T)-f(T_0))$, where f(T) can be derived by an expression obtained by Manoogian and Woolley. Since the oscillator strength is proportional to the number of oscillators per

volume, we approximate the contributions from Nb antisite defects and from the dopant X (X = Mg, Zn) by $A_{Nb_{Li,i}c_{Nb_{Li}}}/(\lambda_{0,i}^{-2} - \lambda^{-2})$ and $A_{X,i}c_{X}/(\lambda_{0,i}^{-2} - \lambda^{-2})$, respectively. For a detailed derivation the reader is referred to Schlarb and Betzler.³ The generalized Sellmeier equation is given by

$$n_i^2 = \frac{A_0, i + A_{\text{Nb}_{\text{Li}}, i} c_{\text{NB}_{\text{Li}}} + A_{X, i} c_X}{\left(\lambda_{0, i} + \mu_{0, i} F\right)^{-2} - \lambda^{-2}} - A_{\text{IR}, i}^2 + A_{\text{UV}}$$
(1)

with

$$\begin{split} c_{\text{Nb}_{\text{Li}}} &= \begin{cases} \frac{2}{3}(50 - c_{\text{Li}}) - c_{\text{X}}/\alpha_{\text{X}} & \text{for } c_{\text{X}} < \alpha_{\text{X}} \frac{2}{3}(50 - c_{\text{Li}}) \\ 0 & \text{for } c_{\text{X}} \ge \alpha_{\text{X}} \frac{2}{3}(50 - c_{\text{Li}}) \end{cases} \\ \alpha_{\text{Mg}} &= 5.0; \alpha_{\text{Zn}} = 6.5; \\ F &= f(T) - f(T_0), T_0 = 24.5^{\circ}\text{C}; \\ f(T) &= (T + 273)^2 + 4.0238 \times 10^5 [\coth\left(\frac{261.6}{T + 273}\right) - 1]. \end{cases}$$

 $c_{\rm X}$ is the dopant concentration given in mol% XO (X = Mg, Zn), $c_{\rm Li}$ denotes the initial Li content, i.e., the ratio [Li₂O]/([Nb₂O₅]+[Li₂O]) extrapolated to undoped material (measured in mol% Li₂O). The parameters $\alpha_{\rm Mg}$ and $\alpha_{\rm Zn}$ (which are proportional to the so-called threshold concentration) were estimated from empirical results reported by Schmidt *et al.*¹¹ and Volk *et al.*² for Mg- and Zn-doped material, respectively. The wavelength λ is given in nm, T in °C, and i=e denotes the extraordinary, i=o the ordinary light polarization.

The parameters for equation (1) were calculated by a fit to our measured refractive index data for undoped and Mg- or Zn-doped lithium niobate and temperature dependent literature data. 12,7 The standard deviation was $\Delta n = 2 \times 10^{-3}$. Numerical results for the parameters are listed in Table I.

Table I
Parameters of the generalized Sellmeier equation. For the definition see Eq. (1) in the text.

n_o	n_e
$\lambda_{0,o} = 223.219$ $\mu_{0,o} = 1.1082 \times 10^{-6}$ $A_{0,o} = 4.5312 \times 10^{-5}$ $A_{Nb_{1i},o} = -7.2320 \times 10^{-8}$ $A_{Mg,o} = -7.3548 \times 10^{-8}$ $A_{Za,o} = 6.7963 \times 10^{-8}$ $A_{IR,o} = 3.6340 \times 10^{-8}$ $A_{UV} = 2.6613$	$\lambda_{0,e}$ = 218.203 $\mu_{0,e}$ = 6.4047 × 10 ⁻⁶ $A_{0,e}$ = 3.9466 × 10 ⁻⁵ $A_{Nb_{Ll}}$, e = 11.8635 × 10 ⁻⁷ $A_{Mg,e}$ = 7.2643 × 10 ⁻⁸ $A_{Za,e}$ = 1.9221 × 10 ⁻⁷ $A_{IR,e}$ = 3.0998 × 10 ⁻⁸ A_{UV} = 2.6613

The generalized Sellmeier equation allows us to derive a simple relation between the phase matching temperature T_{PM} for colinear noncritical type I second harmonic generation and the internal phase matching angle θ_{PM} measured at room temperature.

Defining $\hat{n}_i = n_i(c_{\text{Li}}, c_{\text{X}}, \lambda, T_0)$ and $\tilde{n}_i = n_i(c_{\text{Li}}, c_{\text{X}}, \lambda/2, T_0)$ as the refractive indices for the fundamental (λ) and harmonic ($\lambda/2$) wavelengths at room temperature, respectively, the temperature phase matching condition $n_o(c_{\text{Li}}, c_{\text{X}}, \lambda, T_{\text{PM}}) = n_e(c_{\text{Li}}, c_{\text{X}}, \lambda/2, T_{\text{PM}})$ can be written as

$$(\tilde{n}_e^2 - \hat{n}_o^2) + F \frac{\partial (\tilde{n}_e^2 - \hat{n}_o^2)}{\partial F} |_{F=0} = 0,$$
 (2)

and the parameter F is approximated with coth $x \approx 1/x$ yielding

$$F \approx (T_{\text{PM}} + T_s)^2 - (T_s + T_0)^2; T_s = 1042.$$
 (3)

The angle phase matching condition $\hat{n}_o^{-2} = \tilde{n}_o^{-2} \cos^2 \theta_{PM} + \hat{n}_e^{-2} \sin^2 \theta_{PM}$ can be written as

$$\tilde{n}_e^2 - \hat{n}_o^2 = \cos^2 \theta_{\text{PM}} [\hat{n}_o^2 \frac{\tilde{n}_e^2}{\hat{n}_o^2} - \hat{n}_o^2] \approx \cos^2 \theta_{\text{PM}} [\tilde{n}_e^2 - \tilde{n}_o^2],$$
 (4)

where θ_{PM} denotes—as already mentioned—the direction of propagation in the crystal with respect to the optical axis. A simple calculation yields

$$T_{\text{PM}} = \sqrt{(T_s + T_0)^2 + \gamma(\lambda)\cos^2\theta_{\text{PM}} - T_s}$$

$$\text{with } \gamma(\lambda) \approx (\tilde{n}_o^2 - \hat{n}_e^2) \left[\frac{\partial (\tilde{n}_e^2 - \hat{n}_o^2)}{\partial \tilde{F}} \right]^{-1} |_{F = 0, c_{\text{Li}} = 50, c_{\text{X}} = 0}$$

$$(5)$$

In the wavelength range 1000–1200 nm the parameter $\gamma(\lambda)$ varies between 2.8×10^6 and 3.2×10^6 , we therefore neglect the dispersion and use $\gamma \approx 3 \times 10^6$. In Figure 1 the relation given in Eq. (5) is depicted and compared with experimental data. The results for a variety of crystal compositions and fundamental wavelengths show that the relation is a good approximation for temperatures up to at least 250°C, a maximum deviation of about 20°C is observed. Eq. (5) therefore allows to estimate the phase matching temperature by simply measuring the more easily accessible phase matching angle.

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